BERYLLIUM, 1000 ug/mL (0.10% w/v)

1. Product Identification

**Synonyms:** Atomic Absorption Standard  
**CAS No.:** Not applicable to mixtures.  
**Molecular Weight:** Not applicable to mixtures.  
**Chemical Formula:** Not applicable. (mostly water)  
**Product Codes:** 6921

2. Composition/Information on Ingredients

<table>
<thead>
<tr>
<th>Ingredient</th>
<th>CAS No</th>
<th>Percent</th>
</tr>
</thead>
<tbody>
<tr>
<td>Beryllium</td>
<td>7440-41-7</td>
<td>&lt; 1%</td>
</tr>
<tr>
<td>Hydrogen Chloride</td>
<td>7647-01-0</td>
<td>1 - 2%</td>
</tr>
<tr>
<td>Water</td>
<td>7732-18-5</td>
<td>97 - 98%</td>
</tr>
</tbody>
</table>
3. Hazards Identification

Emergency Overview

DANGER! CORROSIVE. LIQUID AND MIST CAUSE SEVERE BURNS TO ALL BODY TISSUE. MAY BE FATAL IF SWALLOWED OR INHALED. VAPOR IRRITATING TO EYES AND RESPIRATORY TRACT. INHALATION MAY CAUSE LUNG DAMAGE. CANCER HAZARD. CAN CAUSE CANCER. Risk of cancer depends on duration and level of exposure.

J.T. Baker SAF-T-DATA™ Ratings (Provided here for your convenience)

Health Rating: 3 - Severe (Cancer Causing)
Flammability Rating: 0 - None
Reactivity Rating: 2 - Moderate
Contact Rating: 3 - Severe (Corrosive)
Lab Protective Equip: GOGGLES & SHIELD; LAB COAT & APRON; VENT HOOD; PROPER GLOVES
Storage Color Code: White (Corrosive)

Potential Health Effects

Health hazards given on this data sheet apply to concentrated solutions of hydrochloric acid. Hazards of dilute solutions may be reduced, depending upon the concentration. Degree of hazard for these reduced concentrations is not currently addressed in the available literature.

Inhalation:
Corrosive! Inhalation of vapors can cause coughing, choking, inflammation of the nose, throat, and upper respiratory tract, and in severe cases, pulmonary edema, circulatory failure, and death. Inhalation of beryllium and beryllium compounds is associated with an increase risk of lung cancer.

Ingestion:
Corrosive! Swallowing hydrochloric acid can cause immediate pain and burns of the mouth, throat, esophagus and gastrointestinal tract. May cause nausea, vomiting, and diarrhea, and in severe cases, death. Beryllium is poorly absorbed from the gut and has a low oral toxicity.

Skin Contact:
Corrosive! Can cause redness, pain, and severe skin burns. Concentrated solutions cause deep ulcers and discolor skin. Beryllium may gain entrance into superficial cuts and abrasions and cause dermatitis.

Eye Contact:
Corrosive! Vapors are irritating and may cause damage to the eyes. Contact may cause severe burns and permanent eye damage.

Chronic Exposure:
Long-term exposure to concentrated vapors may cause erosion of teeth. Long term exposures
seldom occur due to the corrosive properties of the acid. There is some evidence that beryllium and beryllium compounds cause lung cancer in humans.

Aggravation of Pre-existing Conditions:
Persons with pre-existing skin disorders or eye disease may be more susceptible to the effects of this substance.

4. First Aid Measures

First aid procedures given apply to concentrated solutions. Exposures to dilute solutions may not require these extensive first aid procedures.

Inhalation:
Remove to fresh air. If not breathing, give artificial respiration. If breathing is difficult, give oxygen. Get medical attention immediately.

Ingestion:
If swallowed, DO NOT INDUCE VOMITING. Give large quantities of water. Never give anything by mouth to an unconscious person. Get medical attention immediately.

Skin Contact:
Immediately flush skin with plenty of water for at least 15 minutes while removing contaminated clothing and shoes. Get medical attention immediately. Wash clothing before reuse. Thoroughly clean shoes before reuse.

Eye Contact:
Immediately flush eyes with plenty of water for at least 15 minutes, lifting lower and upper eyelids occasionally. Get medical attention immediately.

5. Fire Fighting Measures

Fire:
Not considered to be a fire hazard.

Explosion:
Not considered to be an explosion hazard. Contact of concentrated solutions with most metals causes formation of flammable and explosive hydrogen gas.

Fire Extinguishing Media:
Use any means suitable for extinguishing surrounding fire.

Special Information:
Structural firefighter’s protective clothing is ineffective for fires involving hydrochloric acid. Stay away from ends of tanks. Cool tanks with water spray until well after fire is out.

6. Accidental Release Measures

Ventilate area of leak or spill. Wear appropriate personal protective equipment as specified in Section 8. Isolate hazard area. Keep unnecessary and unprotected personnel from entering. Contain and recover liquid when possible. Neutralize with alkaline material (soda ash, lime),
then absorb with an inert material (e.g., vermiculite, dry sand, earth), and place in a chemical waste container. Do not use combustible materials, such as saw dust. Do not flush to sewer! US Regulations (CERCLA) require reporting spills and releases to soil, water and air in excess of reportable quantities. The toll free number for the US Coast Guard National Response Center is (800) 424-8802.

J. T. Baker NEUTRASORB® or TEAM® 'Low Na+' acid neutralizers are recommended for spills of this product.

7. Handling and Storage

Store in a cool, dry, ventilated storage area with acid resistant floors and good drainage. Protect from physical damage. Keep out of direct sunlight and away from heat, water, and incompatible materials. Do not wash out container and use it for other purposes. When diluting, the acid should always be added slowly to water and in small amounts. Never use hot water and never add water to the acid. Water added to acid can cause uncontrolled boiling and splashing. When opening metal containers, use non-sparking tools because of the possibility of hydrogen gas being present. Wear special protective equipment (Sec. 8) for maintenance break-in or where exposures may exceed established exposure levels. Wash hands, face, forearms and neck when exiting restricted areas. Shower, dispose of outer clothing, change to clean garments at the end of the day. Avoid cross-contamination of street clothes. Wash hands before eating and do not eat, drink, or smoke in workplace. Containers of this material may be hazardous when empty since they retain product residues (vapors, liquid); observe all warnings and precautions listed for the product.

8. Exposure Controls/Personal Protection

Airborne Exposure Limits:
- OSHA Permissible Exposure Limit (PEL) -
  For Hydrochloric Acid: 5 ppm (Ceiling).
  For Beryllium: 2 ug/m³ (TWA); 5ug/m³ (Ceiling); 25ug/m³, 30-min. (Maximum).

- ACGIH Threshold Limit Value (TLV) -
  For Hydrochloric Acid: 5 ppm (STEL/Ceiling).
  For Beryllium: 2 ug/m³ (TWA), A1 - Confirmed human carcinogen.

Ventilation System:
A system of local and/or general exhaust is recommended to keep employee exposures below the Airborne Exposure Limits. Local exhaust ventilation is generally preferred because it can control the emissions of the contaminant at its source, preventing dispersion of it into the general work area. Please refer to the ACGIH document, *Industrial Ventilation, A Manual of Recommended Practices*, most recent edition, for details.
Personal Respirators (NIOSH Approved):
If the exposure limit is exceeded and engineering controls are not feasible, a full facepiece respirator with an acid gas cartridge may be worn up to 50 times the exposure limit or the maximum use concentration specified by the appropriate regulatory agency or respirator supplier, whichever is lowest. For emergencies or instances where the exposure levels are not known, use a full-facepiece positive-pressure, air-supplied respirator. WARNING: Air purifying respirators do not protect workers in oxygen-deficient atmospheres.

Skin Protection:
Rubber or neoprene gloves and additional protection including impervious boots, apron, or coveralls, as needed in areas of unusual exposure to prevent skin contact.

Eye Protection:
Use chemical safety goggles and/or a full face shield where splashing is possible. Maintain eye wash fountain and quick-drench facilities in work area.

9. Physical and Chemical Properties

Appearance:
Clear, colorless liquid.

Odor:
Slight pungent odor.

Solubility:
Infinitely soluble.

Specific Gravity:
ca. 1.0

pH:
No information found.

% Volatiles by volume @ 21C (70F):
ca. 99

Boiling Point:
ca. 100C (ca. 212F)

Melting Point:
ca. 0C (ca. 32F)

Vapor Density (Air=1):
Essentially the same as water.

Vapor Pressure (mm Hg):
Essentially the same as water.

Evaporation Rate (BuAc=1):
Essentially the same as water.

10. Stability and Reactivity

Stability:
Stable under ordinary conditions of use and storage.
Hazardous Decomposition Products:
Combustion yields toxic fumes of beryllium oxide, hydrogen, and chloride.

Hazardous Polymerization:
Will not occur.

Incompatibilities:
A strong mineral acid, concentrated hydrochloric acid is incompatible with many substances and highly reactive with strong bases, metals, metal oxides, hydroxides, amines, carbonates and other alkaline materials. Incompatible with materials such as cyanides, sulfides, sulfites, and formaldehyde. Beryllium reacts with acids, caustics, chlorinated hydrocarbons, oxidizers, and molten lithium.

Conditions to Avoid:
Heat, direct sunlight, incompatibles.

11. Toxicological Information

Toxicological Data:
Beryllium: No LD50/LC50 information found relating to normal routes of occupational exposure. Investigated as a tumorigen and mutagen. Hydrochloric acid: Inhalation rat LC50: 3124 ppm/1H; Oral rabbit LD50: 900 mg/kg. Investigated as a tumorigen, mutagen, reproductive effector.

Carcinogenicity:
Beryllium and Beryllium Compounds: NIOSH considers this substance to be a potential occupational carcinogen.

---\Cancer Lists\---

<table>
<thead>
<tr>
<th>Ingredient Category</th>
<th>---NTP Carcinogen---</th>
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<tbody>
<tr>
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<td>Known</td>
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<tr>
<td>Beryllium (7440-41-7)</td>
<td>No</td>
</tr>
<tr>
<td>Hydrogen Chloride (7647-01-0)</td>
<td>No</td>
</tr>
<tr>
<td>Water (7732-18-5)</td>
<td>No</td>
</tr>
</tbody>
</table>

12. Ecological Information

Environmental Fate:
For hydrogen chloride: When released into the soil, this material is not expected to biodegrade. When released into the soil, this material may leach into groundwater.

Environmental Toxicity:
For hydrogen chloride: This material is expected to be toxic to aquatic life. For beryllium: The LC50/96-hour values for fish are less than 1 mg/l. The EC50/48-hour values for daphnia are between 1 and 10 mg/l.
13. Disposal Considerations

Whatever cannot be saved for recovery or recycling should be handled as hazardous waste and sent to a RCRA approved waste facility. Processing, use or contamination of this product may change the waste management options. State and local disposal regulations may differ from federal disposal regulations. Dispose of container and unused contents in accordance with federal, state and local requirements.

14. Transport Information

Domestic (Land, D.O.T.)

-----------------------
Proper Shipping Name: CORROSIVE LIQUID, ACIDIC, INORGANIC, N.O.S. (HYDROGEN CHLORIDE)
Hazard Class: 8
UN/NA: UN3264
Packing Group: III
Information reported for product/size: 500ML

International (Water, I.M.O.)

----------------------------
Proper Shipping Name: CORROSIVE LIQUID, ACIDIC, INORGANIC, N.O.S. (HYDROGEN CHLORIDE)
Hazard Class: 8
UN/NA: UN3264
Packing Group: III
Information reported for product/size: 500ML

International (Air, I.C.A.O.)

--------------------------
Proper Shipping Name: CORROSIVE LIQUID, ACIDIC, INORGANIC, N.O.S. (HYDROGEN CHLORIDE)
Hazard Class: 8
UN/NA: UN3264
Packing Group: III
Information reported for product/size: 500ML

15. Regulatory Information

 -----------\Chemical Inventory Status - Part 1\----------------------------------------
 Ingredient TSCA   EC   Japan
 Australia
### Chemical Inventory Status - Part 2

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<th>Ingredient</th>
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<td>Yes</td>
<td>No</td>
<td>Yes</td>
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<tr>
<td>Hydrogen Chloride (7647-01-0)</td>
<td>Yes</td>
<td>Yes</td>
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<td>Water (7732-18-5)</td>
<td>Yes</td>
<td>Yes</td>
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### Federal, State & International Regulations - Part 1

- **SARA 302**

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<td>No</td>
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<tr>
<td>Hydrogen Chloride (7647-01-0)</td>
<td>5000</td>
<td>500*</td>
<td>Yes</td>
<td>No</td>
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<tr>
<td>Water (7732-18-5)</td>
<td>No</td>
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### Federal, State & International Regulations - Part 2

- **RCRA**

<table>
<thead>
<tr>
<th>Ingredient</th>
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<th>-RCRA-261.33</th>
<th>-TSCA-8(d)</th>
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<td>No</td>
<td>No</td>
<td>No</td>
</tr>
</tbody>
</table>

**Chemical Weapons Convention:** No  
**TSCA 12(b):** No  
**CDTA:** No  
**SARA 311/312:** Acute: Yes  
**Chronic:** Yes  
**Fire:** No  
**Pressure:** No  
**Reactivity:** No (Mixture / Liquid)

**WARNING:**
THIS PRODUCT CONTAINS A CHEMICAL(S) KNOWN TO THE STATE OF CALIFORNIA TO CAUSE CANCER.

**Australian Hazchem Code:** None allocated.  
**Poison Schedule:** S6  
**WHMIS:**
This MSDS has been prepared according to the hazard criteria of the Controlled Products Regulations (CPR) and the MSDS contains all of the information required by the CPR.

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16. Other Information
NFPA Ratings: Health: 2 Flammability: 0 Reactivity: 0

Label Hazard Warning:
DANGER! CORROSIVE. LIQUID AND MIST CAUSE SEVERE BURNS TO ALL BODY TISSUE. MAY BE FATAL IF SWALLOWED OR INHALED. VAPOR IRRITATING TO EYES AND RESPIRATORY TRACT. INHALATION MAY CAUSE LUNG DAMAGE. CANCER HAZARD. CAN CAUSE CANCER. Risk of cancer depends on duration and level of exposure.

Label Precautions:
Do not get in eyes, on skin, or on clothing.
Do not breathe vapor or mist.
Use only with adequate ventilation.
Keep container closed.
Wash thoroughly after handling.

Label First Aid:
In case of contact, immediately flush eyes or skin with plenty of water for at least 15 minutes while removing contaminated clothing and shoes. Wash clothing before reuse. If inhaled, remove to fresh air. If not breathing, give artificial respiration. If breathing is difficult, give oxygen. If swallowed, DO NOT INDUCE VOMITING. Give large quantities of water. Never give anything by mouth to an unconscious person. In all cases get medical attention immediately.

Product Use:
Laboratory Reagent.

Revision Information:
No Changes.

Disclaimer:
******************************************************************************
******************************************************************************
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