BERYLLIUM, 1,000 ug/mL or 10,000 ug/mL

1. Product Identification

Synonyms: Single Element Plasma Standard; Atomic Absorption Standard
CAS No.: Not applicable to mixtures.
Molecular Weight: Not applicable to mixtures.
Chemical Formula: Not applicable to mixtures.
Product Codes: 5706, 5720, 6444

2. Composition/Information on Ingredients

<table>
<thead>
<tr>
<th>Ingredient</th>
<th>CAS No</th>
<th>Percent</th>
</tr>
</thead>
<tbody>
<tr>
<td>Beryllium Acetate</td>
<td>543-81-7</td>
<td>&lt; 3%</td>
</tr>
<tr>
<td>Yes</td>
<td></td>
<td></td>
</tr>
<tr>
<td>Nitric Acid</td>
<td>7697-37-2</td>
<td>&lt; 4%</td>
</tr>
<tr>
<td>Yes</td>
<td></td>
<td></td>
</tr>
<tr>
<td>Water</td>
<td>7732-18-5</td>
<td>&gt; 93%</td>
</tr>
<tr>
<td>No</td>
<td></td>
<td></td>
</tr>
</tbody>
</table>
3. Hazards Identification

Emergency Overview

**DANGER! CORROSIVE. LIQUID AND MIST CAUSE SEVERE BURNS TO ALL BODY TISSUE. MAY BE FATAL IF SWALLOWED OR INHALED. VAPOR IRRITATING TO EYES AND RESPIRATORY TRACT. INHALATION MAY CAUSE LUNG AND TOOTH DAMAGE. CANCER HAZARD. CAN CAUSE CANCER. Risk of cancer depends on duration and level of exposure.**

J.T. Baker SAF-T-DATA\textsuperscript{tm} Ratings (Provided here for your convenience)

<table>
<thead>
<tr>
<th>Health Rating</th>
<th>Flammability Rating</th>
<th>Reactivity Rating</th>
<th>Contact Rating</th>
<th>Lab Protective Equip</th>
<th>Storage Color Code</th>
</tr>
</thead>
<tbody>
<tr>
<td>3 - Severe (Cancer Causing)</td>
<td>0 - None</td>
<td>1 - Slight</td>
<td>3 - Severe (Corrosive)</td>
<td>GOGGLES &amp; SHIELD; LAB COAT &amp; APRON; VENT HOOD; PROPER GLOVES</td>
<td>White (Corrosive)</td>
</tr>
</tbody>
</table>

Potential Health Effects

Nitric acid is extremely hazardous; it is corrosive, reactive, an oxidizer, and a poison. The following hazards are for concentrated solutions. Hazards of less concentrated solutions may be reduced. Degree of hazard for reduced concentrations is not currently addressed in the available literature.

**Inhalation:**
Corrosive! Inhalation of vapors can cause coughing, choking, inflammation of the nose, throat, and upper respiratory tract, and in severe cases, pulmonary edema, circulatory failure, and death. Inhalation of beryllium and beryllium compounds is associated with an increase risk of lung cancer.

**Ingestion:**
Corrosive! Swallowing can cause immediate pain and burns of the mouth, throat, esophagus and gastrointestinal tract. May cause nausea, vomiting, and diarrhea, and in severe cases, death.

**Skin Contact:**
Corrosive! Can cause redness, pain, and severe skin burns. Concentrated solutions cause deep ulcers and stain skin a yellow or yellow-brown color.

**Eye Contact:**
Corrosive! Vapors are irritating and may cause damage to the eyes. Contact may cause severe burns and permanent eye damage.

**Chronic Exposure:**
Long-term exposure to concentrated vapors may cause erosion of teeth and lung damage. Long-term exposures seldom occur due to the corrosive properties of the acid. There is some evidence that beryllium and beryllium compounds cause lung cancer in humans.

**Aggravation of Pre-existing Conditions:**
Persons with pre-existing skin disorders, eye disease, or cardiopulmonary diseases may be more susceptible to the effects of this substance.

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### 4. First Aid Measures

Immediate first aid treatment reduces the health effects of this substance.

**Inhalation:**
Remove to fresh air. If not breathing, give artificial respiration. If breathing is difficult, give oxygen. Get medical attention immediately.

**Ingestion:**
If swallowed, **DO NOT INDUCE VOMITING.** Give large quantities of water. Never give anything by mouth to an unconscious person. Get medical attention immediately.

**Skin Contact:**
Immediately flush skin with plenty of water for at least 15 minutes while removing contaminated clothing and shoes. Get medical attention immediately. Wash clothing before reuse. Thoroughly clean shoes before reuse.

**Eye Contact:**
Immediately flush eyes with plenty of water for at least 15 minutes, lifting lower and upper eyelids occasionally. Get medical attention immediately.

---

### 5. Fire Fighting Measures

**Fire:**
Not combustible, but concentrated material is a strong oxidizer and its heat of reaction with reducing agents or combustibles may cause ignition.

**Explosion:**
Concentrated material reacts explosively with combustible organic or readily oxidizable materials such as: alcohols, turpentine, charcoal, organic refuse, metal powder, hydrogen sulfide, etc. Reacts with most metals to release hydrogen gas which can form explosive mixtures with air.

**Fire Extinguishing Media:**
Use any means suitable for extinguishing surrounding fire.

**Special Information:**
In the event of a fire, wear full protective clothing and NIOSH-approved self-contained breathing apparatus with full facepiece operated in the pressure demand or other positive pressure mode.

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### 6. Accidental Release Measures
Ventilate area of leak or spill. Wear appropriate personal protective equipment as specified in Section 8. Isolate hazard area. Keep unnecessary and unprotected personnel from entering. Contain and recover liquid when possible. Neutralize with alkaline material (soda ash, lime), then absorb with an inert material (e.g., vermiculite, dry sand, earth), and place in a chemical waste container. Do not use combustible materials, such as saw dust. Do not flush to sewer! US Regulations (CERCLA) require reporting spills and releases to soil, water and air in excess of reportable quantities. The toll free number for the US Coast Guard National Response Center is (800) 424-8802.

J. T. Baker NEUTRASORB® or TEAM® 'Low Na+' acid neutralizers are recommended for spills of this product.

7. Handling and Storage

Store in a cool, dry, ventilated storage area with acid resistant floors and good drainage. Protect from physical damage. Keep out of direct sunlight and away from heat, water, and incompatible materials. Do not wash out container and use it for other purposes. When diluting, the acid should always be added slowly to water and in small amounts. Never use hot water and never add water to the acid. Water added to acid can cause uncontrolled boiling and splashing. When opening metal containers, use non-sparking tools because of the possibility of hydrogen gas being present. Wear special protective equipment (Sec. 8) for maintenance break-in or where exposures may exceed established exposure levels. Wash hands, face, forearms and neck when exiting restricted areas. Shower, dispose of outer clothing, change to clean garments at the end of the day. Avoid cross-contamination of street clothes. Wash hands before eating and do not eat, drink, or smoke in workplace. Containers of this material may be hazardous when empty since they retain product residues (vapors, liquid); observe all warnings and precautions listed for the product.

8. Exposure Controls/Personal Protection

**Airborne Exposure Limits:**
- OSHA Permissible Exposure Limit (PEL) -
  For Nitric Acid: 2 ppm (TWA).
  For Beryllium and Beryllium Compounds: 2 ug/m3 (TWA); 5ug/m3 (Ceiling); 25ug/m3, 30-min. (Maximum).

- ACGIH Threshold Limit Value (TLV) -
  For Nitric Acid: 2 ppm (TWA); 4 ppm (STEL)
  For Beryllium and Beryllium Compounds: 2 ug/m3 (TWA), A1 - Confirmed human carcinogen.

**Ventilation System:**
A system of local and/or general exhaust is recommended to keep employee exposures below the Airborne Exposure Limits. Local exhaust ventilation is generally preferred because it can control
the emissions of the contaminant at its source, preventing dispersion of it into the general work area. Please refer to the ACGIH document, *Industrial Ventilation, A Manual of Recommended Practices*, most recent edition, for details.

**Personal Respirators (NIOSH Approved):**
If the exposure limit is exceeded and engineering controls are not feasible, wear a supplied air, full-facepiece respirator, airlined hood, or full-facepiece self-contained breathing apparatus. Breathing air quality must meet the requirements of the OSHA respiratory protection standard (29CFR1910.134). Canister-type respirators using sorbents are ineffective.

**Skin Protection:**
Rubber or neoprene gloves and additional protection including impervious boots, apron, or coveralls, as needed in areas of unusual exposure to prevent skin contact.

**Eye Protection:**
Use chemical safety goggles and/or a full face shield where splashing is possible. Maintain eye wash fountain and quick-drench facilities in work area.

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**9. Physical and Chemical Properties**

**Appearance:**
Colorless liquid.

**Odor:**
Odorless.

**Solubility:**
Soluble in water.

**Specific Gravity:**
No information found.

**pH:**
No information found.

**% Volatiles by volume @ 21C (70F):**
ca. 98

**Boiling Point:**
No information found.

**Melting Point:**
No information found.

**Vapor Density (Air=1):**
Not applicable.

**Vapor Pressure (mm Hg):**
Not applicable.

**Evaporation Rate (BuAc=1):**
No information found.

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**10. Stability and Reactivity**
**Stability:**
Stable under ordinary conditions of use and storage.

**Hazardous Decomposition Products:**
When heated to decomposition, emits toxic fumes of nitrogen oxide, beryllium oxide and hydrogen nitrate.

**Hazardous Polymerization:**
Will not occur.

**Incompatibilities:**
A dangerously powerful oxidizing agent, concentrated nitric acid is incompatible with most substances, especially strong bases, metallic powders, carbides, hydrogen sulfide, turpentine, and combustible organics.

**Conditions to Avoid:**
Heat, incompatibles.

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**11. Toxicological Information**

**Toxicological Data:**
No LD50/LC50 information found relating to normal routes of occupational exposure. Beryllium Acetate: Investigated as a tumorigen. Nitric Acid: Investigated as a mutagen, reproductive effector.

**Carcinogenicity:**
Beryllium and Beryllium Compounds: NIOSH considers this substance to be a potential occupational carcinogen.

<table>
<thead>
<tr>
<th>Ingredient Category</th>
<th>---NTP Carcinogen---</th>
<th>IARC</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>Known</td>
<td>Anticipated</td>
</tr>
<tr>
<td>Beryllium Acetate (543-81-7)</td>
<td>No</td>
<td>Yes</td>
</tr>
<tr>
<td>Nitric Acid (7697-37-2)</td>
<td>No</td>
<td>No</td>
</tr>
<tr>
<td>Water (7732-18-5)</td>
<td>No</td>
<td>No</td>
</tr>
</tbody>
</table>

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**12. Ecological Information**

**Environmental Fate:**
No information found.

**Environmental Toxicity:**
For beryllium: The LC50/96-hour values for fish are less than 1 mg/l. The EC50/48-hour values for daphnia are between 1 and 10 mg/l.

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**13. Disposal Considerations**
Whatever cannot be saved for recovery or recycling should be managed in an appropriate and approved waste disposal facility. Processing, use or contamination of this product may change the waste management options. State and local disposal regulations may differ from federal disposal regulations. Dispose of container and unused contents in accordance with federal, state and local requirements.

14. Transport Information

Domestic (Land, D.O.T.)
-----------------------
Proper Shipping Name: CORROSIVE LIQUID, ACIDIC, INORGANIC, N.O.S. (NITRIC ACID)
Hazard Class: 8
UN/NA: UN3264
Packing Group: III
Information reported for product/size: 500ML

International (Water, I.M.O.)
-----------------------------
Proper Shipping Name: CORROSIVE LIQUID, ACIDIC, INORGANIC, N.O.S. (NITRIC ACID)
Hazard Class: 8
UN/NA: UN3264
Packing Group: III
Information reported for product/size: 500ML

International (Air, I.C.A.O.)
-----------------------------
Proper Shipping Name: CORROSIVE LIQUID, ACIDIC, INORGANIC, N.O.S. (NITRIC ACID)
Hazard Class: 8
UN/NA: UN3264
Packing Group: III
Information reported for product/size: 500ML

15. Regulatory Information

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Chemical Inventory Status - Part 1---

<table>
<thead>
<tr>
<th>Ingredient</th>
<th>TSCA</th>
<th>EC</th>
<th>Japan</th>
</tr>
</thead>
<tbody>
<tr>
<td>Beryllium Acetate (543-81-7)</td>
<td>Exempt</td>
<td>Yes</td>
<td>No</td>
</tr>
<tr>
<td>Nitric Acid (7697-37-2)</td>
<td>Yes</td>
<td>Yes</td>
<td>Yes</td>
</tr>
</tbody>
</table>
Water (7732-18-5) Yes Yes Yes Yes

--- Chemical Inventory Status - Part 2 ---

Ingredient | Korea | DSL | NDSDL | Phil.
--- | --- | --- | --- | ---
Beryllium Acetate (543-81-7) | No | No | No | No
Nitric Acid (7697-37-2) | Yes | Yes | No | Yes
Water (7732-18-5) | Yes | Yes | No | Yes

--- Federal, State & International Regulations - Part 1 ---

Ingredient | RQ | TPQ | List | Chemical Catg.
--- | --- | --- | --- | ---
Beryllium Acetate (543-81-7) | No | No | No | Beryllium co
Nitric Acid (7697-37-2) | 1000 | 1000 | Yes | No
Water (7732-18-5) | No | No | No | No

--- Federal, State & International Regulations - Part 2 ---

Ingredient | CERCLA | -RCRA- | -TSCA-
--- | --- | --- | ---
Beryllium Acetate (543-81-7) | No | No | No
Nitric Acid (7697-37-2) | 1000 | No | No
Water (7732-18-5) | No | No | No

Chemical Weapons Convention: No TSCA 12(b): No CDTA: No SARA 311/312: Acute: Yes Chronic: Yes Fire: No Pressure: No Reactivity: No (Mixture / Liquid)

WARNING:
THIS PRODUCT CONTAINS A CHEMICAL(S) KNOWN TO THE STATE OF CALIFORNIA TO CAUSE CANCER.

Australian Hazchem Code: None allocated.
Poison Schedule: S5
WHMIS:
This MSDS has been prepared according to the hazard criteria of the Controlled Products Regulations (CPR) and the MSDS contains all of the information required by the CPR.

16. Other Information

NFPA Ratings: Health: 3 Flammability: 0 Reactivity: 0
Label Hazard Warning:
DANGER! CORROSIVE. LIQUID AND MIST CAUSE SEVERE BURNS TO ALL BODY
TISSUE. MAY BE FATAL IF SWALLOWED OR INHALED. VAPOR IRRITATING TO EYES AND RESPIRATORY TRACT. INHALATION MAY CAUSE LUNG AND TOOTH DAMAGE. CANCER HAZARD. CAN CAUSE CANCER. Risk of cancer depends on duration and level of exposure.

**Label Precautions:**
- Do not get in eyes, on skin, or on clothing.
- Do not breathe vapor or mist.
- Use only with adequate ventilation.
- Wash thoroughly after handling.
- Keep container closed.

INTENDED FOR R & D USE ONLY.
NOT ON THE TSCA INVENTORY.

**Label First Aid:**
In case of contact, immediately flush eyes or skin with plenty of water for at least 15 minutes while removing contaminated clothing and shoes. Wash clothing before reuse. If inhaled, remove to fresh air. If not breathing, give artificial respiration. If breathing is difficult, give oxygen. If swallowed, DO NOT INDUCE VOMITING. Give large quantities of water. Never give anything by mouth to an unconscious person. In all cases get medical attention immediately.

**Product Use:**
Research and Development Use Only.

**Revision Information:**
- No Changes.

**Disclaimer:**
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